Phase Changes Gizmo Answer Key

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Heat and phase changes

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Answers will vary. [Air pressure is lower at higher elevations, which causes water to boil at a lower temperature. The cooler boiling water cooks the spaghetti more slowly.] Gizmo Warm-up In the Phase Changes Gizmo, select Micro view and set the Ice volume to 50 cm 3. Notice the nitrogen (), oxygen (), and water molecules.

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Get the Gizmo ready: Activity A: Phase changes. Click Reset () and select Macro view. Set the Water temperature to 10 C. Set the Ice

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volume to 0 cc. Question: How is temperature related to phase changes? 1. Predict: Based on your prior knowledge, predict the following: 0 A. At what temperature will water change from a liquid to a solid (freeze)? _____ 0 B. At what temperature will water change from a solid to a liquid (melt)?

phasechangesse | Ice | Freezing

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Phase changes. Get the Gizmo ready: Click Reset and select Macro view. Set the Water temperature to 10 °C. Set the Ice volume to 0 cc. Question: How is temperature related to phase changes?

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Gizmo Warm-up In the Phase Changes GizmoTM, select the Micro view and set the Ice volume to 50 cc. Click Play () and observe molecules in the solid (ice), liquid (water), and gas (air) phases. 1. In which phase (s) are the molecules held rigidly together?

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Bubbles are invisible forms of water. Heat energy makes water into gas bubbles as phase change occurs. The heat lifts the gas bubbles into the air and pops at the surface of the water releasing water into the air. Why does phase change occur?

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Get the Gizmo ready: Click Reset, and select the Micro view. Set Ice volume to 0 cc. Set Add/remove heat energy to 0 J/s. Question: Why do phase changes occur? 1. Compare: Set the Water temperature to 0 °C and click Play. Observe the water molecules. Click Reset, set the Water temperature to 100 °C, and click Play again. What do you notice?

Get the Gizmo ready: Activity B: Reset Micro view ...

Phase Changes. Explore the relationship between molecular motion, temperature, and phase changes. Compare the molecular structure of solids, liquids, and gases. Graph temperature changes as ice is melted and water is boiled. Find the effect of altitude on phase changes. The starting temperature, ice volume, altitude, and rate of heating or cooling can be adjusted.

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