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Name CHAPTER 3 TEST continued Date Class FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9, If a particular compound is composed of elements A and B, the ratio of the mass of B to the mass of A will always be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

~~San Ramon Valley High School~~

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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~~3 Atoms: The Building Blocks of Matter~~

Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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number of protons and neutrons in the nucleus of an isotope.

~~3 Atoms: The Building Blocks of Matter~~

Modern Chemistry 105 Chapter Test Name _____ Class _____ Date _____ Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7.

A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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minutes to answer all of the questions in this Standardized Test Preparation Answer the following items on a separate piece of paper.
MULTIPLE CHOICE I. In the modern periodic table, elements are arranged according to

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