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Introduction to Chemistry 1.1 The
Stories of Two Chemicals Ozone
Layer, atmosphere, ozone
formation, chlorofluorocarbons,
CFC ' s 1.2 Chemistry and Matter
Chemistry Central Science

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Chapter Summaries – Chemistry
Matter and Change

Chapter 1 Section 2 MATTER and
Its Properties Matter – Anything
that has MASS and VOLUME.

ATOMS and MOLECULES the
basic building blocks of matter.

ATOM-- the smallest unit of an

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element that still maintains the
properties that element.

ELEMENT – a pure substance
made up of only one type of atom.

MATTER and CHANGE Chapter 1
Section 1

Chemistry: Matter and Change

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Chapter 6 Study Guide for Content
Mastery . Name Date CHAPTER
Class STUDY GUIDE FOR
CONTENT MASTERY Section 6.3
Periodic Trends In your textbook,
read about atomic radius and ionic
radius. Circle the letter of the
choice that best completes the

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statement or answers the question.

1. Atomic radii cannot be measured ...

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CHAPTER 3 SOLUTIONS

MANUAL Matter—Properties and
Changes Matter—Properties and
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Chapter 3 35 Section 3.1

Properties of Matter pages 70 – 75

Problem-Solving Lab 1. Explain

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why the flow of a compressed gas must be controlled for practical and safe use.

Matter—Properties and
Changes Matter—Properties and
Changes

Chapter 1 VIDEODISC Cosmic

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Effect, Movie Ozone Studies, Movie
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ROM Chemistry: Matter and
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Chemistry, Demonstration The
following multimedia for this

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Chapter 1: Introduction to
Chemistry

Chemistry: Matter and Change

Teacher Guide and Answers 7

Study Guide - Chapter 6 – The

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Periodic Table and Periodic Law
Section 6.1 Development of the
Modern Periodic Table 1. octaves
2. eight 3. nine 4. accepted 5.
Dmitri Mendeleev 6. atomic mass
7. elements 8. atomic number 9.
Henry Moseley 10. protons 11.
periodic law 12. properties 13. 14

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Ch 6 Study Guide answers

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Chapter 5 25 Electrons in Atoms

Reviewing Vocabulary Match the
definition in Column A with the

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term in Column B. Column A
Column B 1. The set of
frequencies of the electromagnetic
waves emitted by the atoms of an
element 2. The minimum amount
of energy that can be lost or
gained by ...

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Chapter Assessment

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key. some numbers used in
chemistry can be q... the study of
energy changes (that is, energy
produced or absor... A system that
can exchange neither energy nor

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Change. extensive property.
intensive property. extensive

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properties (examples) intensive
property (examples) a property
that depends on the amount of
matter in a sample. a property that
depends on the type of matter in a
sample not.... mass, volume,
length, weight.

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Chapter 11 31 number of moles of
known substance C H C H C H O O
O CO CO CO H O H O H O number
of moles of unknown substance O

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CO H O C H CO H O C H O H O C
H O CO use mole ratio 5 mol O 2 1
mol C 3 H 8 3 mol CO 2 1 mol C 3
H 8 4 mol H 2 O 1 mol C 3 H 8 1
mol C 3 H 8 5 mol O 2

MATH SKILLS TRANSPARENCY
MASTER 16

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A “ chemical change ” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a

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chemical change. Properties of
Compounds Quite different
properties than their component
elements.

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