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(energy, matter,
elements, reactions)

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PLAY. Molarity. a unit describing
CONCENTRATION of
a solution. solutions.
homogeneous mixture
of 2 or more pure
substances. a solution
is composed of.
solvent & solute.
solvent. in greater
quantity than solute,
typically a liquid but
not always. solute.

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PLAY. A solution is a.

homogeneous

mixture. The first step

of solvation is to

_____ thermal

energy, whereas the

mixing process _____

it. to absorb; release.

A student desires to

remove the color of

naphthalene (moth

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balls), a nonpolar aromatic hydrocarbon, from his laboratory coat. What solvent should he ...

Answers

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Electronegativities:

Classify the following bonds as nonpolar covalent, polar covalent, or ionic. If a bond is polar covalent, identify which atom has the partial negative charge and which has the partial positive charge. If a bond is ionic, identify which atom

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Chapter 12 Liquids:
Condensation,
Evaporation, and
Dynamic ...

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calorimetry. heat
capacity. specific
heat.

thermochemistry. the
study of heat flow and
measurement. the
amount of heat

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needed to raise the
temperature of an
object. heat capacity
of 1 gram of a
substance. the study
of the changes in heat
in chemical reactions.

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Chapter 12 □ States of

Matter Section 12.1

Gases 1. motion 2. a.

small b. forces c.

random d. elastic;

kinetic 3. $KE = \frac{1}{2}mv^2$

4. Temperature 5.

true 6. true 7. false 8.

true 9. true 10. false

11. true 12. false 13.

a 14. a 15. d 16. d 17.

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Compare and contrast
sublimation and
evaporation. In both

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processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

States of Matter
CHAPTER STUDY

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GUIDE FOR CONTENT

MASTERY Section

12.2 Stoichiometric

Calculations

In your textbook, read about mole-to-mole conversion.

Read the following passage and then solve the

problems. In the equation that follows each problem, write in the space provided

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the mole ratio that can
be used to solve the
problem.

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and Carboxylic Acids.
The next ones in line
are the aldehydes,

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ketones and
carboxylic acids. The
IUPAC names, to
start with, the chapter
explores their
preparation methods,
the key reactions,
characteristic tests,
and other properties.
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Matter and Change.

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biological, physical.

Name the two

categories of science.

the science of

different kinds of

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matter and how that
matter can change.

Define Chemistry.

organic. the study of
carbon based
compounds.

inorganic. the study of
non-

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