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balanced?...

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Here is the

balanced symbol

equation:  $2\text{Cu} +$

$\text{O}_2 \rightarrow 2\text{CuO}$ . You

can see that now



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there are two  
copper atoms and  
two oxygen atoms  
on each side.

This matches  
what happens in  
the reaction.

Two atoms of ...

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A compound is found to have a molecular mass of 90 atomic mass units and simplest formula of  $C_2H_5O$ .

The molecular formula of the substance is:

\*\*Use atomic masses of  $C = 12$  amu,  $H = 1$  amu,  $O = 16$  amu\*\* A.

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C 3 H 6 O 3 B. C  
4 H 26 O C. C 4  
H 10 O 2 D. C 5  
H 14 O

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Practice Test  
Questions  
The chemical  
formula of a  
compound shows  
how many of each  
type of atom

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Join together to  
make the units  
which make up  
the compound.

For example, in  
iron sulfide  
every iron atom  
is joined to  
one...

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$2\text{NH}_3$ . Reactants

Products. N : 2

N :  $1 \times 2 = 2$ . H :

$2 \times 3 = 6$  H :  $3 \times 2 =$

6. NOTE : When

you add a

coefficient in

front of a

compound, all

the elements in

the compound

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The chemical  
equation has the  
products on the  
right side,  
while the  
reactants are

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written on the  
left side. Both  
of them are  
separated by an  
arrow. For  
instance,  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$   
denotes that  
there are four  
atoms of  
hydrogen and 2  
atoms of oxygen  
on both sides of  
the equation.

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The amount of  
reactants must  
be equal to the  
amount of  
products.

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in space,  
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although there  
is A formula  
used to indicate  
not just the  
range of atoms.  
Molecular  
formulas  
describe the  
particular  
number and  
variety of atoms

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within a  
molecule of a  
compound. A  
molecular

formula isn't a  
compound name,  
and no words are  
included by it.

Writing Chemical  
Formulas

Worksheet Answer  
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To set up an



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Equations Gizmo,  
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type the

chemical  
formulas into  
the text boxes  
of the Gizmo.  
First, type in  
“H<sub>2</sub>+O<sub>2</sub>” in the  
Reactants box  
and “H<sub>2</sub>O” in the  
Products box.  
This...

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(ANSWER ...

The chemical  
equation for  
this reaction is  
written as:  $2\text{H}_2$   
 $+ \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

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Other symbols are used to indicate the starting materials, or reactants, which by convention are written on the left side of the equation, and the final compounds, or products, which are written on

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The theory

explains

chemical

formulas and has

four major

points: • All

matter is made

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of atoms that cannot be divided or destroyed. • All atoms of a given element are the same in mass and properties. • New compounds are formed when two or more elements combine.

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Solution for  
(3b-101) Balance  
the following  
equations (be  
sure to first  
convert any  
chemical names  
into formulas).  
Show all steps  
for balancing or

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write a brief...

Answered:

(3b-101) Balance  
the following... |  
bartleby

Worksheet 1

Writing And

Balancing

Chemical

Equations. The  
numbers of N and  
O atoms on  
either side from

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the equations are now equal, and so the equation is usually

balanced. Check Your Learning

Write a balanced equation for the decomposition of ammonium nitrate to form

molecular nitrogen,  
molecular

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oxygen, and water.

Chemical Reaction Worksheet #1 Writing And Balancing ...

1. These structures break in reactants. \_ \_ \_ s 2. Equal numbers of each kind of atom on



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each side of the  
arrow. — — a —  
— — — 4. Change  
in size, shape,  
or state ...

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A chemical  
equation  
describes what

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happens in a  
chemical  
reaction. The  
equation

identifies the  
reactants  
(starting  
materials) and  
products  
(resulting  
substances), the  
formulas of the  
participants,  
the phases of

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the participants  
(solid, liquid,  
gas), the  
direction of the  
chemical  
reaction, and  
the amount of  
each substance.

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