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a) Ethane in staggered conformation has C<sub>3</sub>-rotational axis passing through the C-C bond. It has three C<sub>2</sub>-rotational axes perpendicular to the principal C<sub>3</sub>-rotational axis and through the midpoint of the C-C bond. It has not any h-plane. But it has 3 d-planes containing two carbons and anti two hydrogens. Hence the point group of ethane in the staggered conformation will be D<sub>3d</sub>.

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law summarizes what happens, e.g., law of conservation of mass in a chemical reaction or the ideal gas law,  $PV = nRT$ . A theory (model) is an attempt to explain why something happens.

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