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Because a smaller amount of NH<sub>3</sub> is produced from the H<sub>2</sub> than from the N<sub>2</sub>, the amount of H<sub>2</sub> must be limiting. Thus because the H<sub>2</sub> is the limiting reactant, the amount of NH<sub>3</sub> that can form is 1.65  $\times$  10<sup>3</sup> moles.

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